

Photochemical Ring Expansion of Diazidonaphthalenes: Formation of the First Examples of Azepinoazepines

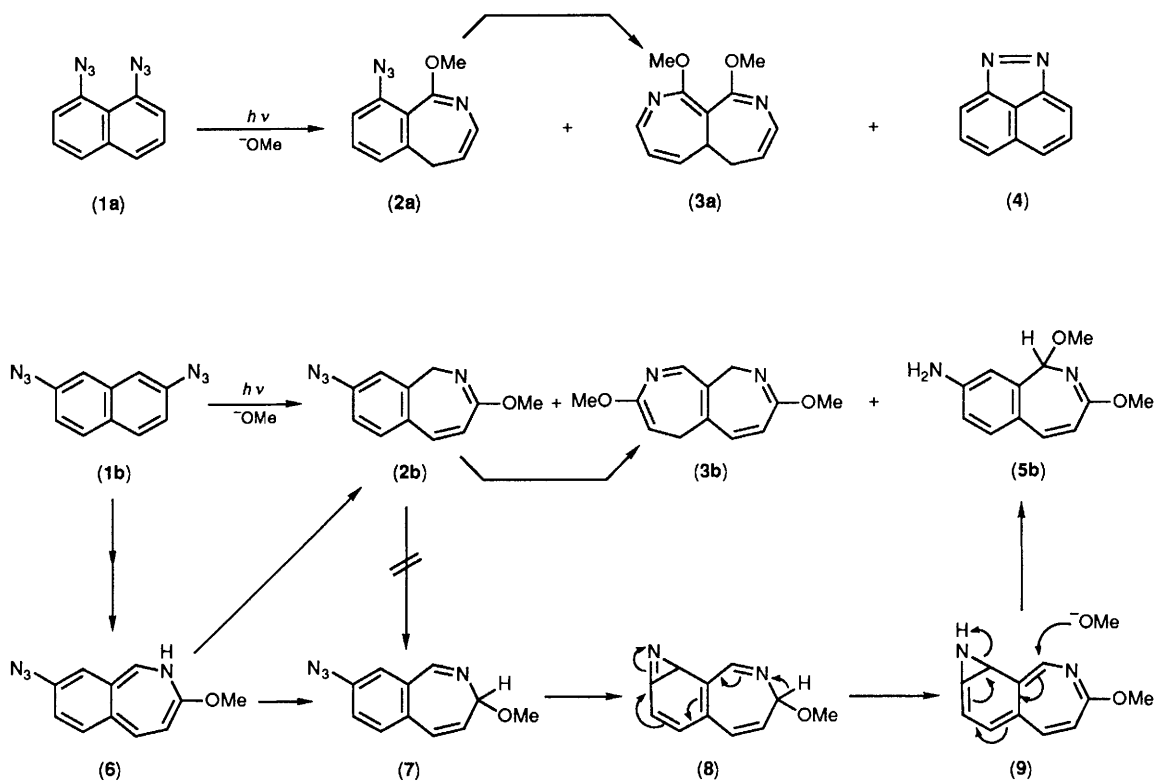
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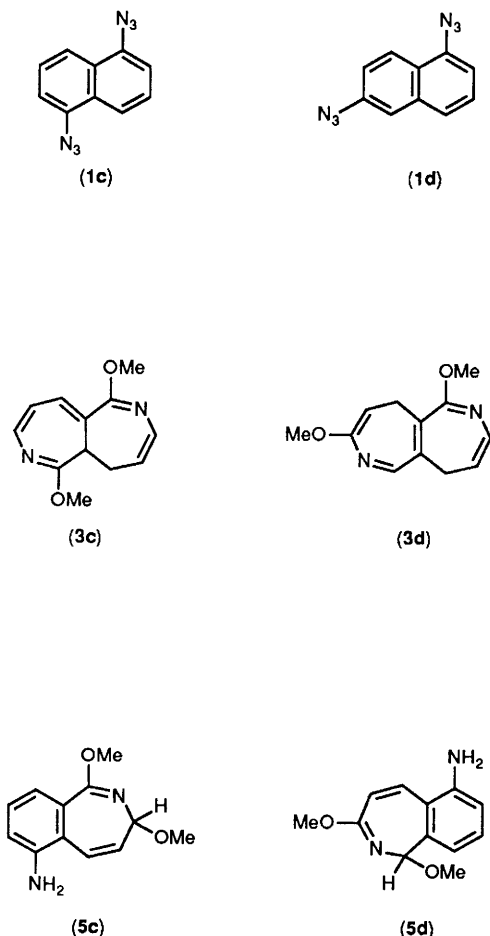
Photolysis of 1,5-, 1,6-, 1,8-, and 2,7-diazidonaphthalenes (**1**) results in ring expansion of both benzene rings to give the corresponding novel azepinoazepines (**3**) *via* the azidobenzazepines (**2**).

There have been several interesting reports on the thermal or photochemical decomposition of diazidonaphthalenes. 1,8-Diazidonaphthalene gives benz[*cd*]indazole,¹ while 1,2-² and 2,3-diazidonaphthalenes³ undergo ring opening or rearrangements to form various nitrile compounds. On the

other hand, phenyl,⁴ pyridyl,⁵ and benzopyridyl⁶ azides are known to undergo ring expansion *via* nitrene and azirine intermediates giving rise to seven-membered nitrogen heterocycles, upon photolysis or thermolysis in the presence of nucleophiles. However, there have been no reports



Scheme 1



concerning the ring expansion of diazidonaphthalenes and, in particular, the simultaneous expansion of both rings of condensed bicyclic compounds. We report here that the diazidonaphthalenes (**1**) underwent photochemical ring expansion of both rings to give the novel heterocycles (**3**) which are the first examples of azepinoazepine ring systems.

Photolysis (400 W, high-pressure Hg lamp; Pyrex filter) of the diazidonaphthalene (**1a–d**) was carried out in methanol-dioxane (1 : 1) containing a large excess of sodium methoxide with ice cooling until almost all the starting azides had been consumed (for 20–40 min).[†] 1,8-Diazidonaphthalene (**1a**) gave the 9-azido-5H-2-benzazepine (**2a**) (15–20%) and the 1,10-dimethoxyazepino[3,4-c]azepine (**3a**) (10–15%),[‡] together with benz[*cd*]indazole (**4**) (ca. 40%). 2,7-Diazidonaphthalene (**1b**) also afforded two ring-expansion products

[†] Irradiation times longer than 20–40 min resulted in a decrease in the amounts of products, especially (**3**).

[‡] Satisfactory elemental analyses and spectroscopic data were obtained for all new compounds reported. *Selected ¹H NMR data* (CDCl₃): (**2a**) (m.p. 45 °C) δ 2.92 (2H, d, 5-H₂), 5.30 (1H, dt, 4-H), and 6.52 (1H, d, 3-H), *J*_{3,4} 7 and *J*_{4,5} 8 Hz; (**2b**) (m.p. 93–94 °C) δ 4.20 (2H, s, 1-H₂), 6.20 (1H, d, 4-H), and 7.08 (1H, d, 5-H), *J*_{4,5} 12 Hz; (**3a**) (m.p. 56–58 °C) δ 2.68 (1H, dd, 5-H), 2.86 (1H, d, 5a-H), 3.1–3.4 (1H, m, 5-H'), 3.66 and 3.72 (each 3H, s, OMe), 5.12 (1H, ddd, 4-H), 5.8–6.0 (2H, m, 6- and 7-H), 6.24 (1H, dd, 3-H), and 6.86 (1H, dd, 8-H), *J*_{3,4} 8, *J*_{3,5'} 2, *J*_{4,5} 8, *J*_{4,5'} 6, *J*_{5,5'} 14, *J*_{5',5a} 2, *J*_{6,7} 8, *J*_{6,8} 2, and *J*_{7,8} 8 Hz; (**3b**) (m.p. 68–69 °C) δ 2.56 (2H, d, 6-H₂), 3.62 and 3.66 (each 3H, s, OMe), 3.88 (2H, s, 1-H₂), 5.36 (1H, t, 7-H), 5.64 (1H, d, 4-H), 6.40 (1H, d, 5-H), and 6.80 (1H, s, 10-H), *J*_{4,5} 11 and *J*_{6,7} 7 Hz. The locations of the sp³ carbon atoms in the azepine rings of (**3**) were confirmed by ¹³C NMR spectroscopy: (**3a**) δ 32.36 (t, C-5) and 53.60 (d, C-5a); (**3b**) δ 33.65 (t, C-6) and 51.24 (t, C-1); (**3c**) δ 33.59 (t, C-5) and 54.12 (d, C-5a); (**3d**) δ 31.18 and 33.36 (each t, C-5 and -10).

(2b) and (3b) in similar yields, as well as the unexpected aminobenzazepine (5b) (40–50%) having two methoxy groups in the azepine ring. § The compounds (2) isolated were converted to (3) by further irradiation for 5–10 min under conditions similar to those for (1) in 30–40% yields.

A possible mechanism for the photolysis of (1b) is shown in Scheme 1. The unstable 2*H*-2-benzazepine (6) initially formed from (1b) may undergo a [1,7] hydrogen shift in the azepine ring to yield two tautomers (2b) and (7). The 1*H*-isomer (2b) gives only the azepinoazepine (3b) and no aminobenzazepine (5b), while the unisolable 3*H*-isomer (7) might be converted into (5b) via the intermediates (8) and (9).

Similarly, 1,5- (1c) and 1,6-diazidonaphthalene (1d) gave the azepinoazepines (3c) (15–20%) and (3d) (20–25%), and aminobenzazepines (5c) (5–10%) and (5d) (15–20%), respectively, as well as the corresponding azidobenzazepines (2). In the case of the unsymmetrical azide (1d), of course, two kinds of azidobenzazepines were obtained.

The compounds (3) reported are the first examples of azepinoazepines and considered to be aza-analogues of

dihydroheptalenes. Heptalenes are typical non-aromatic cyclic compounds with twelve π electrons and can be prepared by oxidation of dihydroheptalenes,⁷ but their aza-analogues are unknown. However, attempts to convert (3) into fully unsaturated diazaheptalenes have been unsuccessful.

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§ Selected data for (5b): m.p. 113–115 °C; ¹H NMR δ 3.64 and 3.76 (each 3H, s, OMe), 4.04 (2H, br., NH₂), 4.90 (1H, s, 1-H), 6.26 (1H, d, 4-H), 6.68 (1H, dd, 7-H), 7.24 (1H, d, 5-H), and 7.1–7.2 (2H, m, 6- and 9-H), $J_{4,5}$ 12, $J_{6,7}$ 8, and $J_{7,9}$ 2 Hz; ¹³C NMR (sp³-C) δ 51.7 (q), 53.3 (q), and 88.6 (d). The structures of (5) were also confirmed by chemical studies. Details including the results of several reactions of (2) and (3) will be published elsewhere.
